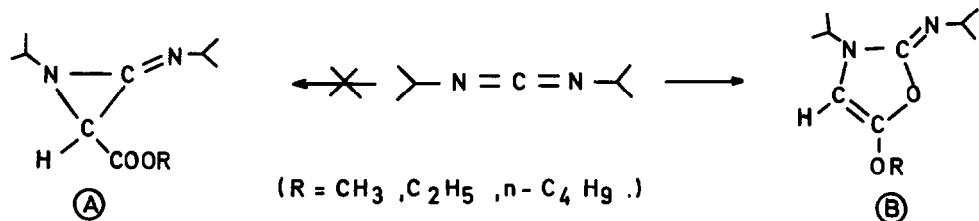


NOVEL IMINOOXAZOLINES FROM REACTIONS OF
DIAZOACETATES WITH CARBODIIMIDES (A REVISION)¹

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In a previous article²⁾, four of us reported that alkyl diazoacetates react with *N,N'*-diisopropylcarbodiimide in the presence of transition metal salts (copper triflate or rhodium(II) acetate) to give 1-isopropyl-2-alkoxycarbonyl-3-isopropylimino-aziridine (A).



A revision of the structure³⁾ and a reinterpretation of the spectral data in the light of an X-ray diffraction analysis⁴⁾ showed that our compounds are in fact 2-isopropylimino-3-isopropyl-5-alkoxy-4-oxazolines (B).

The crystallographical data⁴⁾ as well as the results reported below were obtained on highly purified B (R=CH₃, m.p.=51.5 - 52°C).

The i.r. spectrum (KBr) is best explained by attributing the 1725 and the 1665 cm⁻¹ bands respectively to the exocyclic (C=N-) imino group and the endocyclic (C=C) bond.

The ¹H n.m.r. spectrum in CDCl₃ (100 MHz, TMS as reference) shows singlets at δ(ppm) = 5.33 (4H oxazoline) and 3.71 (CH₃-O), two heptuplets (CH isopropyl) at 4.14 (³J=6.7 Hz) (N-CH) and 3.64 (³J=6.3 Hz) (C=N-CH) and two doublets (CH₃ isopropyl) at 1.16 (³J=6.7 Hz) and 1.03 (³J=6.3 Hz).

The ^{13}C n.m.r. spectrum in CDCl_3 (TMS as reference) gives signals at δ (ppm) = 150.9 and 147.0 (ring C^2 and C^5), 84.6 ($^1\text{J}=194.1$ Hz) (C^4H), 46.7 ($^1\text{J}=131.0$ Hz) and 45.2 ($^1\text{J}=138.5$ Hz) (CH isopropyl), 25.0 ($^1\text{J}=122.7$ Hz) and 20.7 ($^1\text{J}=125.0$ Hz) (CH_3 isopropyl) and 58.6 ($^1\text{J}=143.7$ Hz) ($\text{O}-\text{CH}_3$).

The mass spectrum of (B, $\text{R}=\text{CH}_3$) exhibits signals at M/e (relative intensity) (70 eV) : 198 (25, molecular peak), 183 (9, $\text{M}-\text{CH}_3$), 169 (8), 156 (11, $\text{M}-\text{C}_3\text{H}_6$), 155 (14, $\text{M}-\text{C}_3\text{H}_7$), 141 (23, $\text{M}-\text{C}_3\text{H}_7\text{N}$), 114 (25, $\text{M}-2\text{C}_3\text{H}_6$), 113 (100, $\text{M}-\text{C}_3\text{H}_7\text{NCO}$), 99 (40), 97 (20), 82 (22), 70 (50, $\text{M}-\text{C}_3\text{H}_7\text{NCO}-\text{C}_3\text{H}_7$), 58 (50, $\text{CH}_2\text{N}_2\text{O}$ from 114), 56 (8, $\text{CH}=\text{COCH}_3$ from 114) and 43 (69, C_3H_7^+).

The oxazolines (B) are weak bases, soluble in dilute acids, and can be titrated potentiometrically in methanol with HClO_4 0.1 N in dioxane (equivalent weight for (B), $\text{R}=\text{CH}_3$: found 210; calcd. 198)

They react with picric acid in aqueous solution to give a monoadduct. For example, (B), $\text{R}=\text{CH}_3$, gives the corresponding salt as yellow needles after crystallization from benzene. Mp : 125°C. Calcd for $\text{C}_{16}\text{H}_{21}\text{N}_5\text{O}_9$: C, 44.95; H, 4.95; N, 16.4 %. Found : C, 45.0; H, 4.9; N, 16.3 %. M/e (70 eV) : 427 (calcd 427).

The mechanism of the formation of (B) involves most probably a 1,3-dipolar addition of the carbalkoxycarbene onto a C=N- group of the cumulene, and is in fact similar to the cycloaddition of carbalkoxycarbenes onto nitriles⁵⁾.

Conclusion : The reported reaction, which illustrates the difficulty to obtain exocyclic methylene aziridines, is a new straightforward method for the preparation of novel 4-oxazolines which are not easily accessible by other methods.

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- 1) Part I of the series, part II will be devoted to the chemical properties of the reported oxazolines.
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